Preliminary communication

The reaction of allene with β -diketonatorhodium(I) complexes: formation of π -allylic derivatives of rhodium(III)

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Considerable interest has been focused recently on the reactions between allene and rhodium complexes¹⁻⁸. Various types of reactions have been observed, such as formation of complexes of rhodium(I) with allene¹⁻⁵ or oligomers of allene⁵⁻⁷, catalytic cyclooligomerization⁶⁻⁸, and linear polymerization¹. Most of the studies were concerned with rhodium monohalide complexes.

We have investigated the reaction of allene with $(acac)Rh(C_2H_4)_2$ and $(DBM)Rh(C_2H_4)_2^*$ and have observed the formation of unusual complexes of rhodium-(III) in which a linear allene tetramer is bonded to the metal by two π -allylic bonds. We report here the preparation and structure of these complexes.

When allene is bubbled through a suspension of $(DBM)Rh(C_2H_4)_2$ in pentane (260 mg in 8 ml) at room temperature the precipitate readily dissolves to give a clear solution, from which yellow crystals, I, are obtained upon cooling to 0° (yield about 60%). The crystals of I are stable in the air for several days; the elemental analysis is in accord with the formula $(DBM)Rh(C_3H_4)_4$.

Complex I does not react at room temperature either with CO (atmospheric pressure) or with cyclo-1,5-octadiene. Passing H₂ at room temperature through a suspension of I in ethanol gives a saturated hydrocarbon, M^+ at m/e 170, which has been found to be identical with 2,3,6,7-tetramethyloctane⁹.

The NMR spectrum of I is complicated, and has not been fully interpreted. The structure of the complex has been completely determined by X-ray examination of single crystals. Complex I crystallizes in the monoclinic space group $P2_1/n$ with eight moleccles in the unit cell. The main crystal data are: a = 14.50 Å, b = 23.86 Å, c = 12.93 Å, $\beta = 94.65^{\circ}$. The intensities of 4650 independent reflections were evaluated using a Picker automated diffractometer (Mo-K α radiation). The structure was solved by Patterson-Fourier methods and refined by the least squares method. In the asymmetric unit two distinct molecules are present; the same molecular geometry has been found for both with each individual molecule having C_2 symmetry. The final molecular model is represented in Fig. 1.

*acac = acetylacetonato; DBM = 1,3-diphenylpropan-1,3-dionato.



Fig. 1. Molecular model of complex I.

Only the carbon atoms $C_1 C_2 C_3$ and $C'_1 C'_2 C'_3$ are coordinated to rhodium, the Rh- C_i and Rh- C'_i distances being 2.18, 2.21 and 2.11 Å (i = 1, 2, 3); the bond lengths C_1 - C_3, C_2 - C_3 , and C'_1 - C'_3, C'_2 - C'_3 are identical (1.44 Å) and the two groups C_1 - C_2 - C_3 - C_4 , C'_1 - C'_2 - C'_3 - C'_4 are planar. This geometry indicates that the (C_3 H₄)₄ group is bonded to Rh by two allylic bonds and that it is to be formulated as:



From the reaction of allene with $(acac)Rh(C_2H_4)_2$ a yellow compound, II, of formula $(acac)Rh(C_3H_4)_4$ has been obtained as a microcrystalline powder. II gives 2,3,6,7-tetramethyloctane on treatment with H₂, which suggests that its structure is analogous to that of I.

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